



Bio-Nanohybrid Materials

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AIM OF THIS COMMUNICATION

- Overview of procedures to obtain **biopolymer nanocomposites** provided with specific functionality
- Applications of such materials towards advanced devices
- Research results from our own laboratory

ORGANIC-INORGANIC HYBRID MATERIALS

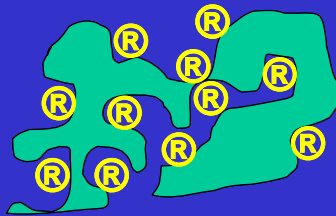
- Grafting of organic groups



EXAMPLES

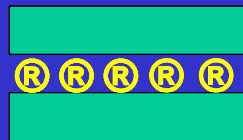
*grafting of organosilanes
on silicic surfaces*

- Sol-gel methods



*self-templating synthesis of
organosilicic compounds*

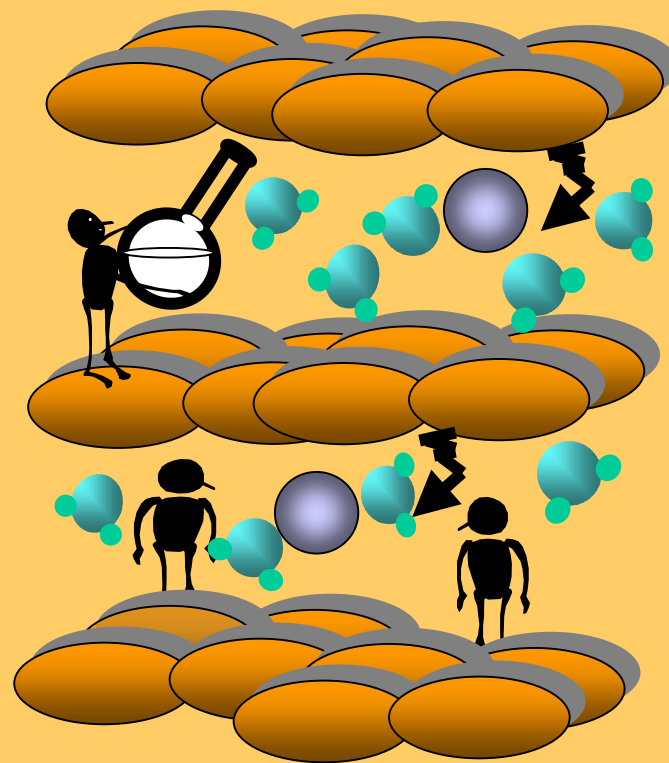
- Intercalation of organic compounds



*polymer & biopolymer
nanocomposites*

Nanostructured Organic-Inorganic Functional materials have to be prepared by a combination of physical and chemical methods following the modus operandi characteristic of the procedures of molecular engineering.

Idealized representation of a “nanochemist” applying methods for deliberate modifications of the functional properties of a 2D solid material. The picture includes some laborers of theoretical formation studying physical processes into the solid.



E. RUIZ-HITZKY “Organic-Inorganic Materials: From Intercalations to Devices”. Chapter 2. In: P. Gómez-Romero, C. Sánchez, eds. Functional Hybrid Materials; Wiley-VCH Verlag GmbH, 2004.

STRUCTURAL & FUNCTIONAL NANOCOMPOSITES

COMPOSITE MATERIALS

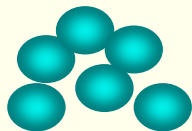
Are solids resulting from the combination of two or more simple materials that develop a **continuous phase** and a **dispersed phase** which together have a set of properties that is essentially different from the components taken separately.

polymer
metal
ceramic...

glass fiber
silica...

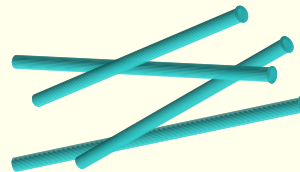
NANOCOMPOSITE MATERIALS

Are composites in which the dispersed phase presents at least one dimension at the nanometric scale.



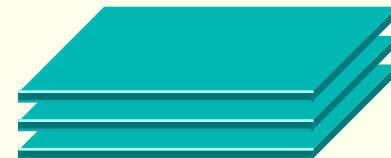
3-nano-D

Silica, C...



2-nano-D

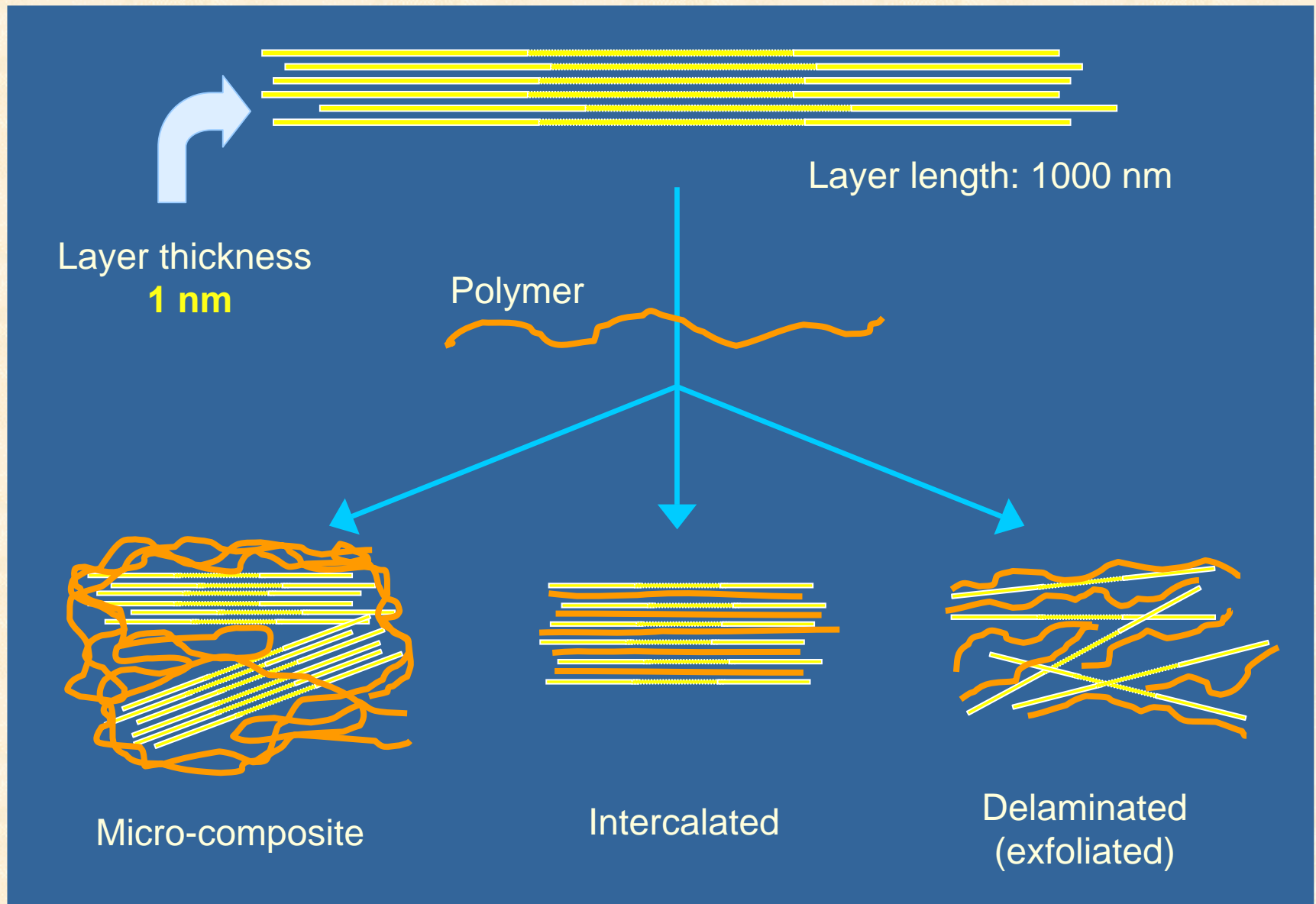
Fibres, tubes...



1-nano-D

Layered solids

HYBRID NANOCOMPOSITES BASED ON ORGANIC POLYMERS AND LAYERED SOLIDS



EXAMPLES OF POLYMER-LAYERED SOLID NANOCOMPOSITES

Possibilities:

- **inorganic host:** non-charged, positively or negatively charged layers, insulator, semiconductor, redox-sites, ...
- **polymers:** insulators, conducting, ionomers, polyelectrolytes...

Examples:

Host	polymer
<u>layered silicates (clays)</u> (montmorillonite, ..)	PVA, PS, PSS, PMMA, PAN, PEG, PEO, PANi, PPy
<u>layered double hydroxides</u> (MgAl, ZnAl, ZnCr,...)	PSS, PVS, PAA, PANi
phosphates & phosphonates (α -ZrP,...)	PPy, PANi,...
metal oxides ($V_2O_5 \cdot nH_2O$, CoO_2 , MoO_3 ,...)	PVP, PPV, PANi, PPy, PTh, PEG, PEO
metal chlorides & oxychlorides ($FeOCl$, α - $RuCl_3$,...)	PVP, PPy, PANi, PTh, PEO
metal chalcogenides (MoS_2 , $NbSe_2$,...)	PEO, PANi
metal phosphochalcogenides ($MnPS_3$, $CdPS_3$,...)	PEO, PANi,
graphite oxide	PEO, PANi

LAYERED SILICATES (CLAYS)

Natural or synthetic phyllosilicates. Type 2:1 charged silicates (montmorillonite, hectorite,..)

TWO tetrahedral layers of M^{IV} (eg. Si) which are coordinated by oxygen anions to M^{III} (eg. Al) in ONE octahedral layer

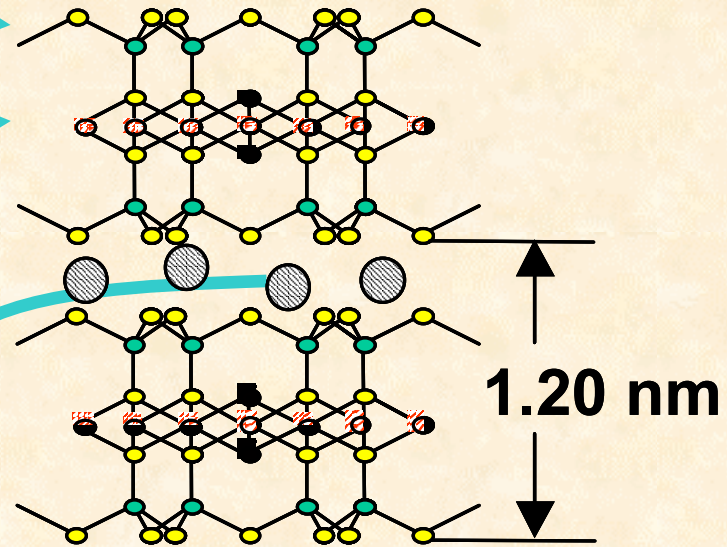
substitution of M^{III} cations by M^{II} cations in the octahedral layer and/or substitution of M^{IV} cations by M^{III} cations in the tetrahedral layer



overall negative charge

a large variety of characteristics due to:

- nature of cations M^{II} , M^{III} and M^{IV}
- nature of the interlayer cation



electrical neutrality is kept by exchangeable interlayer cations

LAYERED DOUBLE HYDROXIDES (LDHs)

LDHs are *synthetic* or natural solids with a layered structure similar to that exhibited by natural $Mg(OH)_2$, brucite.

layers of M^{II} and M^{III} cations which are octahedrally coordinated by six oxygen anions, as hydroxides

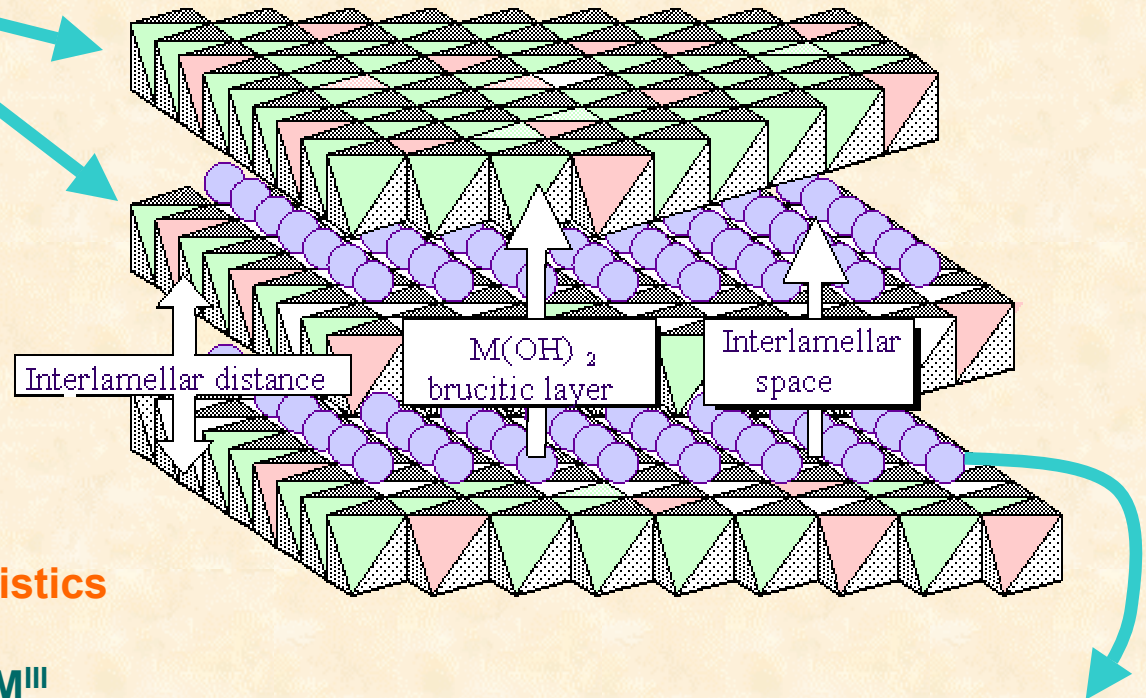
substitution of M^{III} cations into the brucite-like hydroxide layer



overall positive charge

a large variety of characteristics due to:

- nature of cations M^{II} and M^{III}
- nature of the interlayer anion



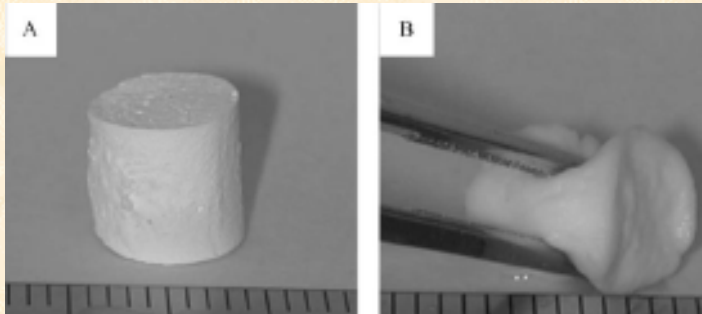
electrical neutrality is kept by exchangeable interlayer anions

Characteristics:

- high charge density in the layers
- anion-exchange properties: ion-exchange capacity 2-3 times greater than clays

Bio-nanocomposites : State of the Art & Examples

Bio-nanohybrids from collagen/alginate and hydroxyapatite: biomaterials for bone tissue engineering and drug delivery:



newly formed bone retained the original size of the bio-nanocomposite implant

(Sotome et al. *Mater. Sci. Engineering C*, 2004, 24, 341-347)

**Negatively charged DNA has been intercalated in a LDH matrix
LDH works as a nonviral vector to transfer the DNA into the cells:**



(Choy, et al. *Angew. Chem. Int. Ed.*, 2000, 39, 4042-4045)

OBJECTIVES

The development of bio-nanocomposites based on the intercalation of natural polysaccharides into layered solids (clays & LDHs)



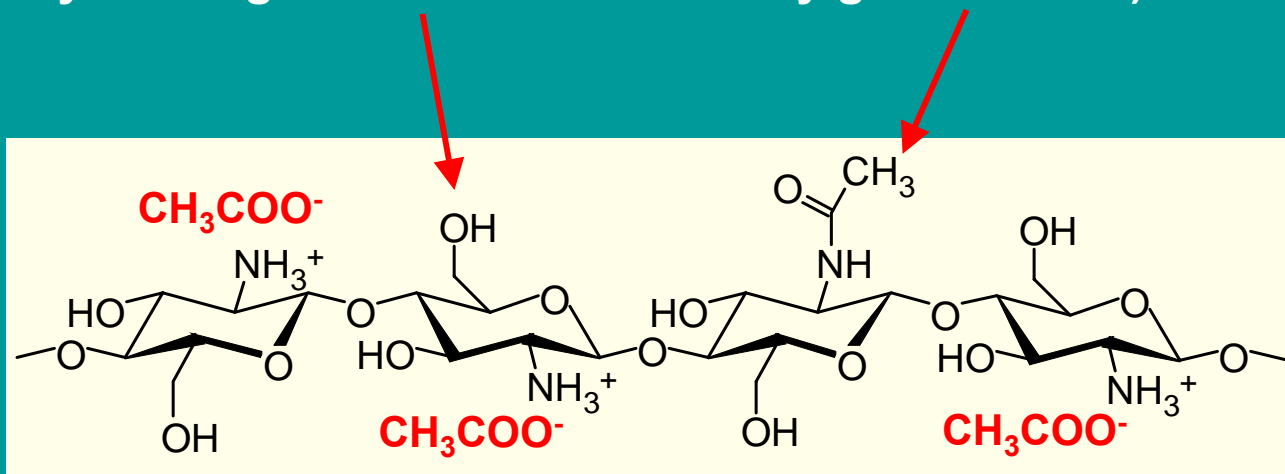
*Reinforcement of biopolymers
Tuneable ion-exchange behaviour*



- Application of the bio-nanocomposites as active phase of potentiometric sensors*
- Controlled release of electrically charged drugs*

CHITOSAN-CLAY BIO-NANOCOMPOSITES

Chitosan: bio-polymer constituted by saccharide units (is a copolymer of glucosamine and N-acetylglucosamine).

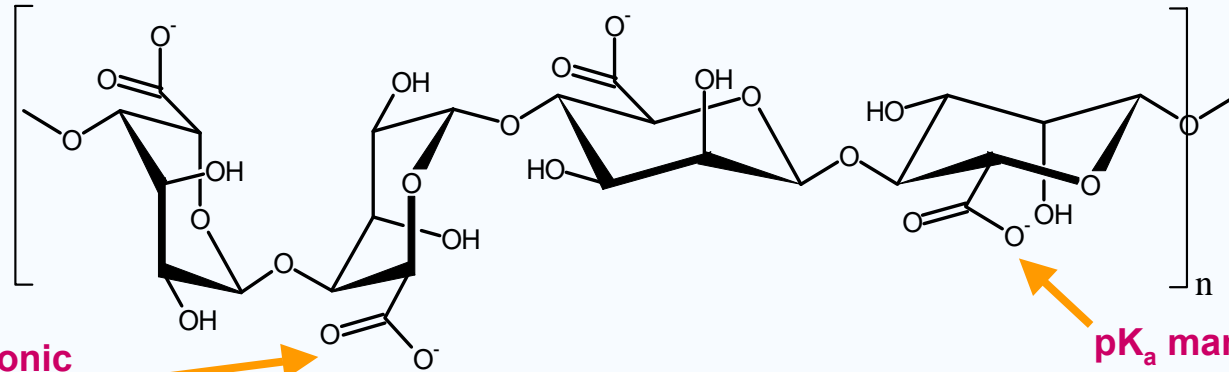


In acid solutions (acetic acid, 1%) as a positively charged polymer ($-\text{NH}_3^+$, $\text{pK}_a = 6.3$), which is adsorbed on smectite clay suspensions.

- Ion-exchange process
- Polymer-salt intercalation

M. DARDER, M. COLILLA, E. RUIZ-HITZKY *Biopolymer-clay nanocomposites based on chitosan intercalated in montmorillonite* Chem. Mater. **15**, 3774-3780 (2003)

ANIONIC POLYSACCHARIDES LDH BIO-NANOCOMPOSITES

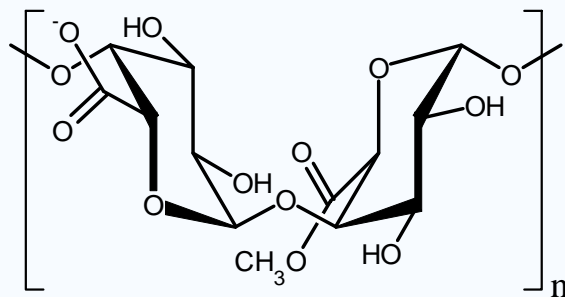


pK_a guluronic acid units = 3.65

pK_a mannuronic acid units = 3.38

Alginate

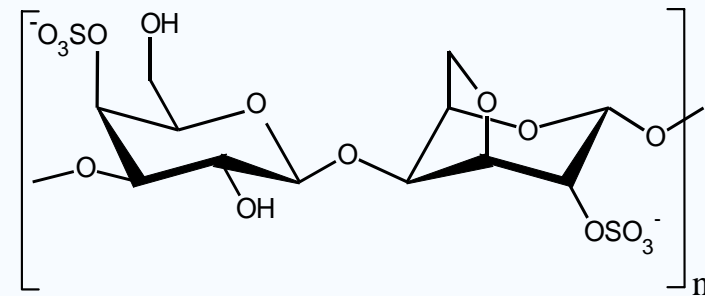
From brown algae



$pK_a = 2.9$

Pectin

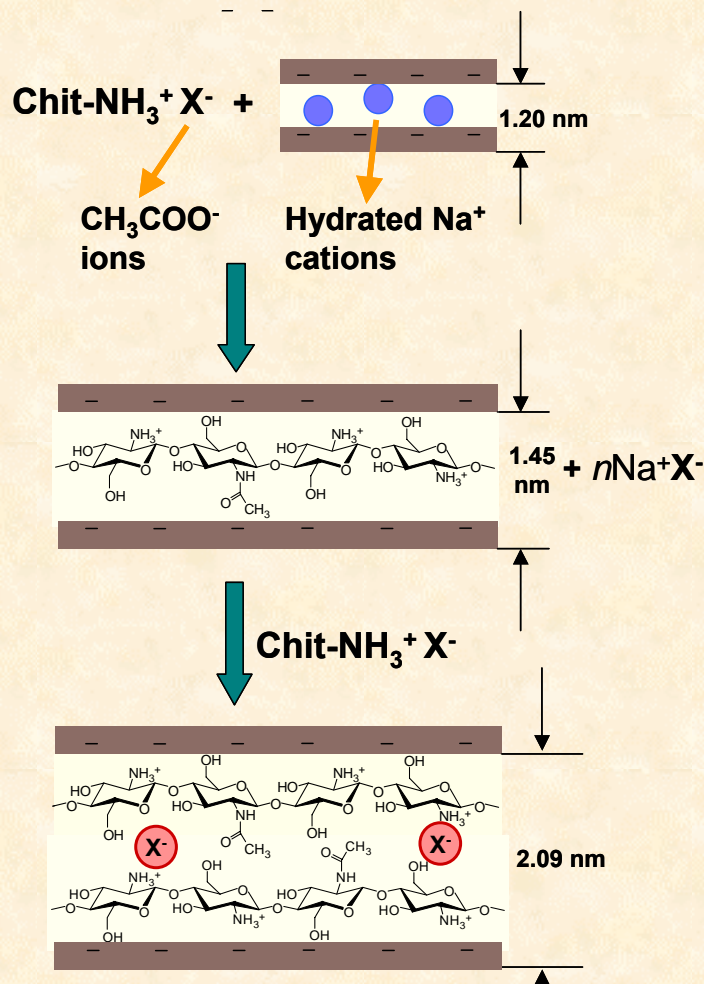
From citrus fruits



iota-Carrageenan

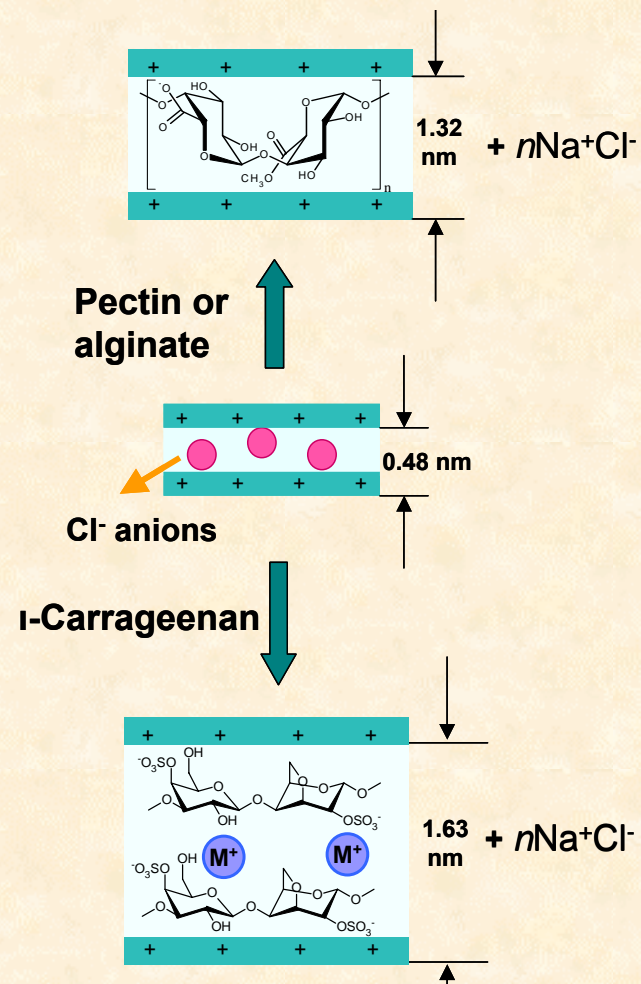
From red algae

chitosan-montmorillonite bio-nanocomposites



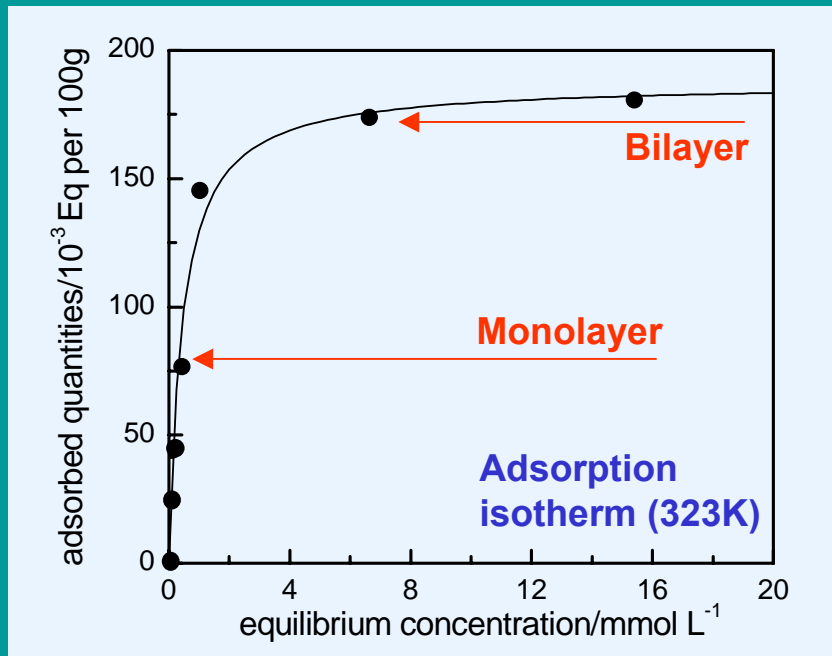
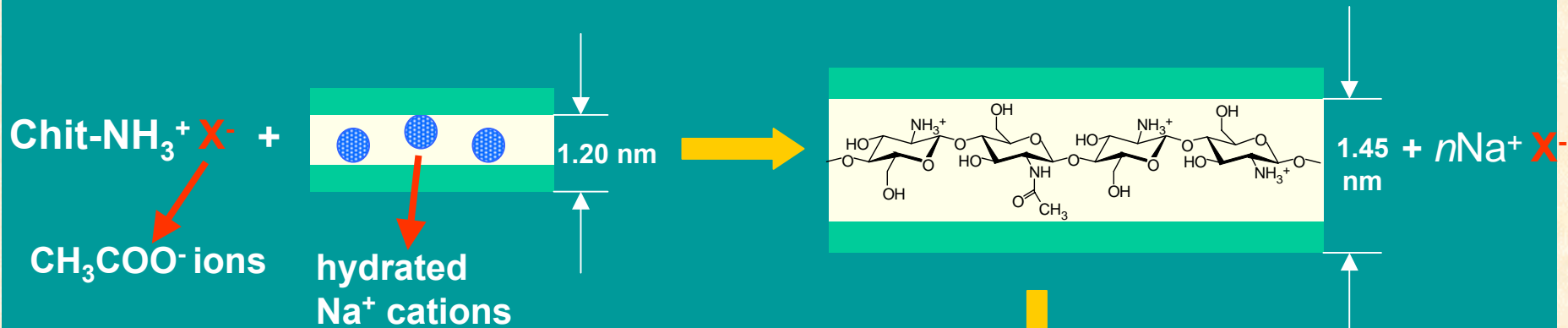
M. DARDER, M. COLILLA, E. RUIZ-HITZKY
*Biopolymer-clay nanocomposites based on
 chitosan intercalated in montmorillonite*
Chem. Mater. 15, 3774-3780 (2003)

anionic polysaccharides- LDH bio-nanocomposites



M. DARDER, M. LOPEZ-BLANCO, P. ARANDA, F.
 LEROUX, E. RUIZ-HITZKY *Bio-nanocomposites
 based on layered double hydroxides*
Chem. Mater. 17, 1969-1977 (2005)

Intercalation of chitosan in Na⁺-montmorillonite

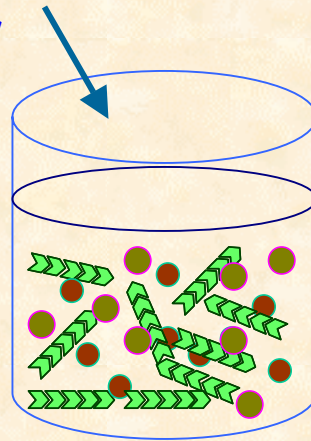


Theoretical anionic exchange capacity of 57 mEq/100g

ONE-POT SYNTHESIS OF BIO-POLYMER/LDH NANOCOMPOSITES

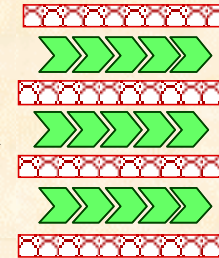
Co-organized assembly or
templating co-precipitation:

LDH precursors
+ polymer

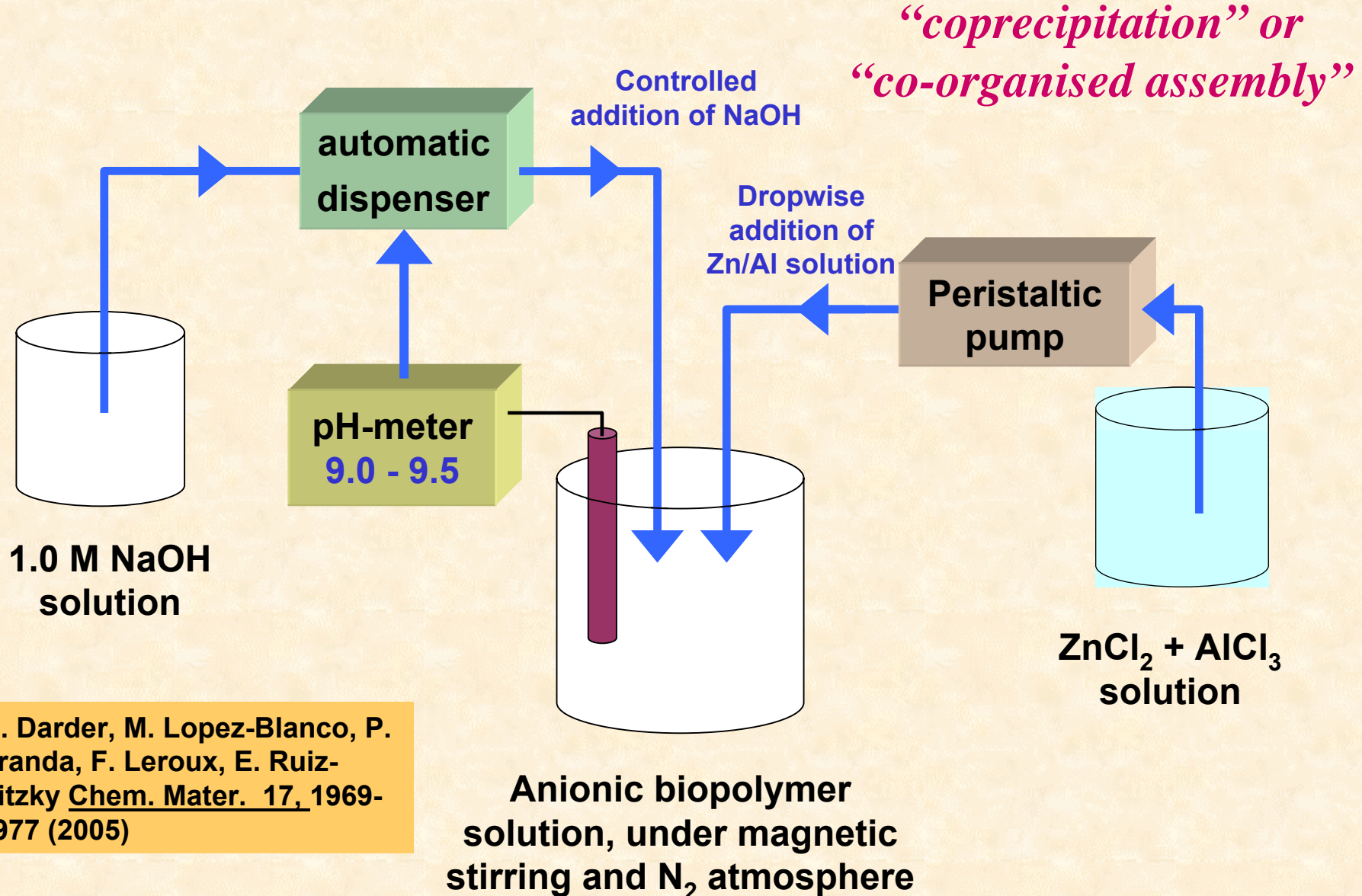


control of pH

Bio-nanocomposite



TEMPLATE SYNTHESIS OF BIOPOLYMER-LDH NANOCOMPOSITES

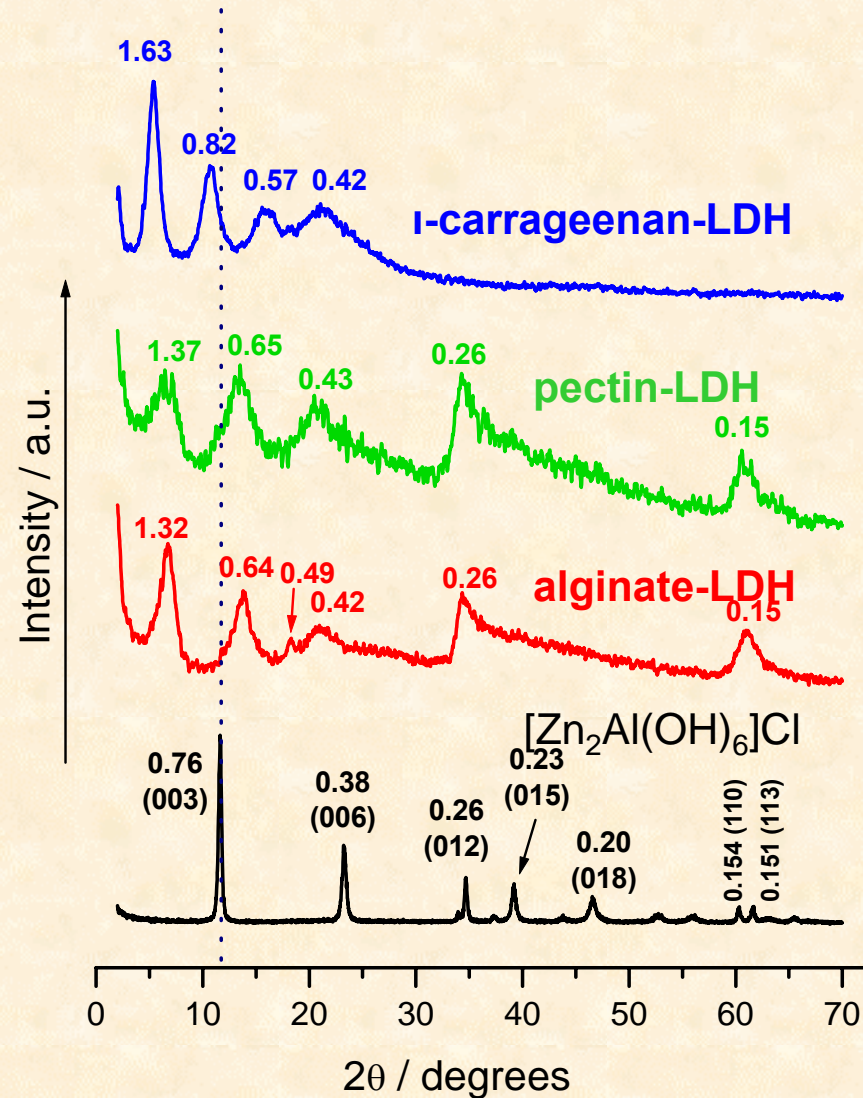
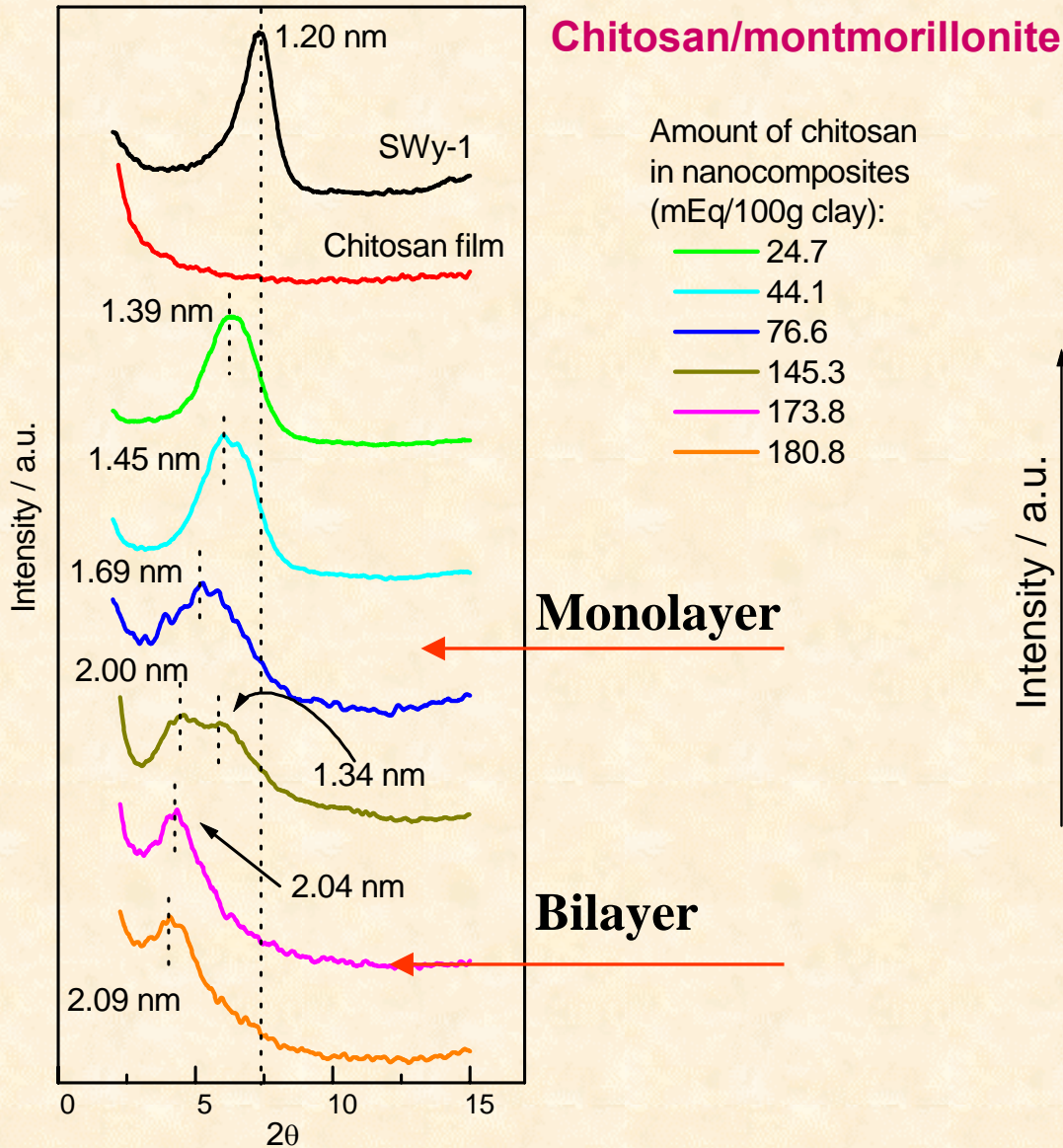


M. Darder, M. Lopez-Blanco, P. Aranda, F. Leroux, E. Ruiz-Hitzky *Chem. Mater.* **17**, 1969-1977 (2005)

CHARACTERIZATION OF THE BIOPOLYMER- NANOCOMPOSITES

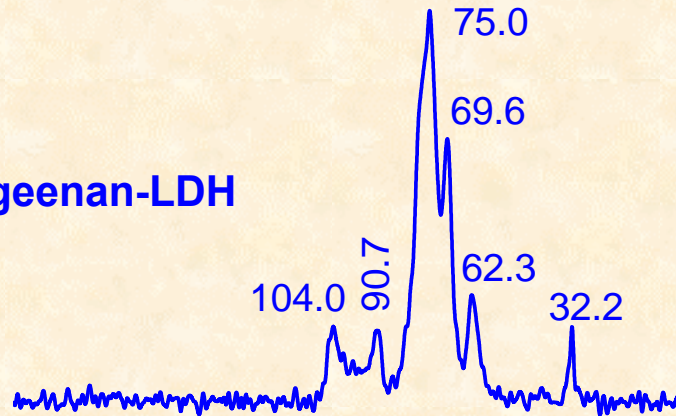
- X-ray diffraction
- FTIR spectroscopy
- Solid state ^{13}C NMR spectroscopy
- Scanning electron microscopy (SEM)
- Thermogravimetry (TG) and differential thermal analysis (DTA)
- Direct potentiometry

X-Ray Diffraction

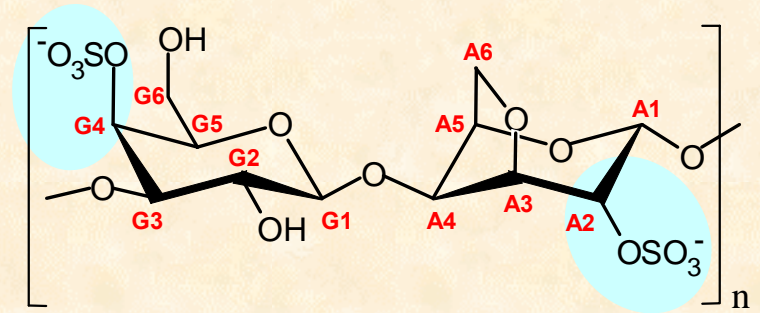
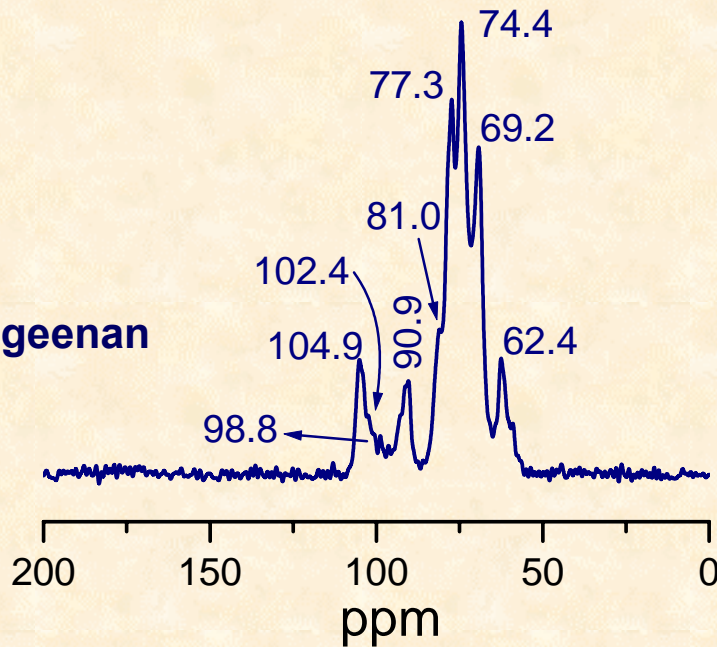


Solid-state ^{13}C NMR spectroscopy

I-carrageenan-LDH



I-carrageenan



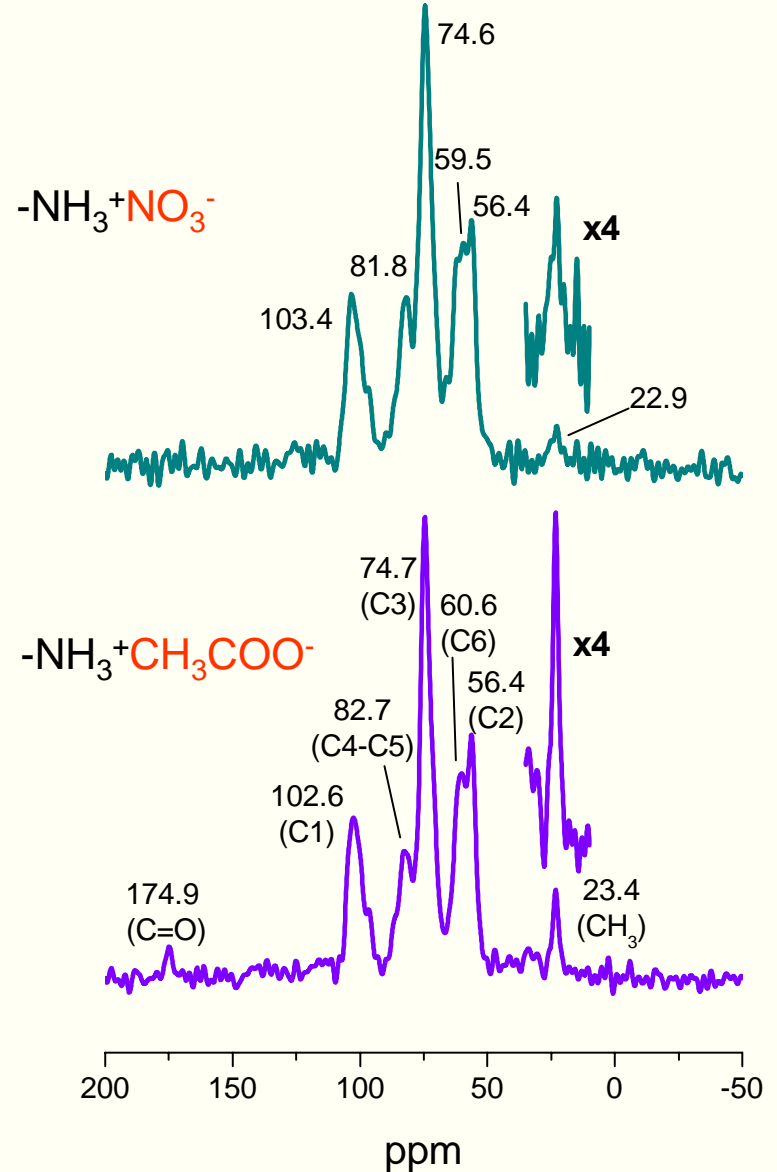
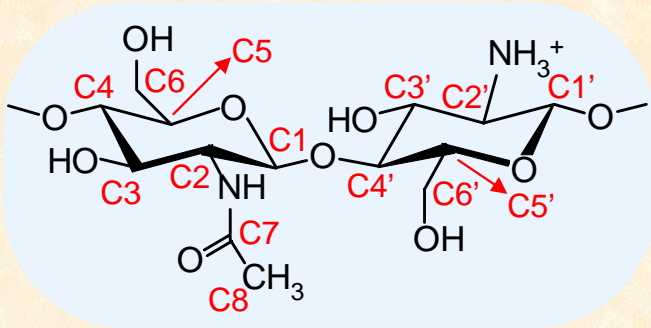
G4: 74.4 ppm \rightarrow 75.0 ppm
A2: 69.2 ppm \rightarrow 69.6 ppm

Chitosan-clay bio-nanocomposite

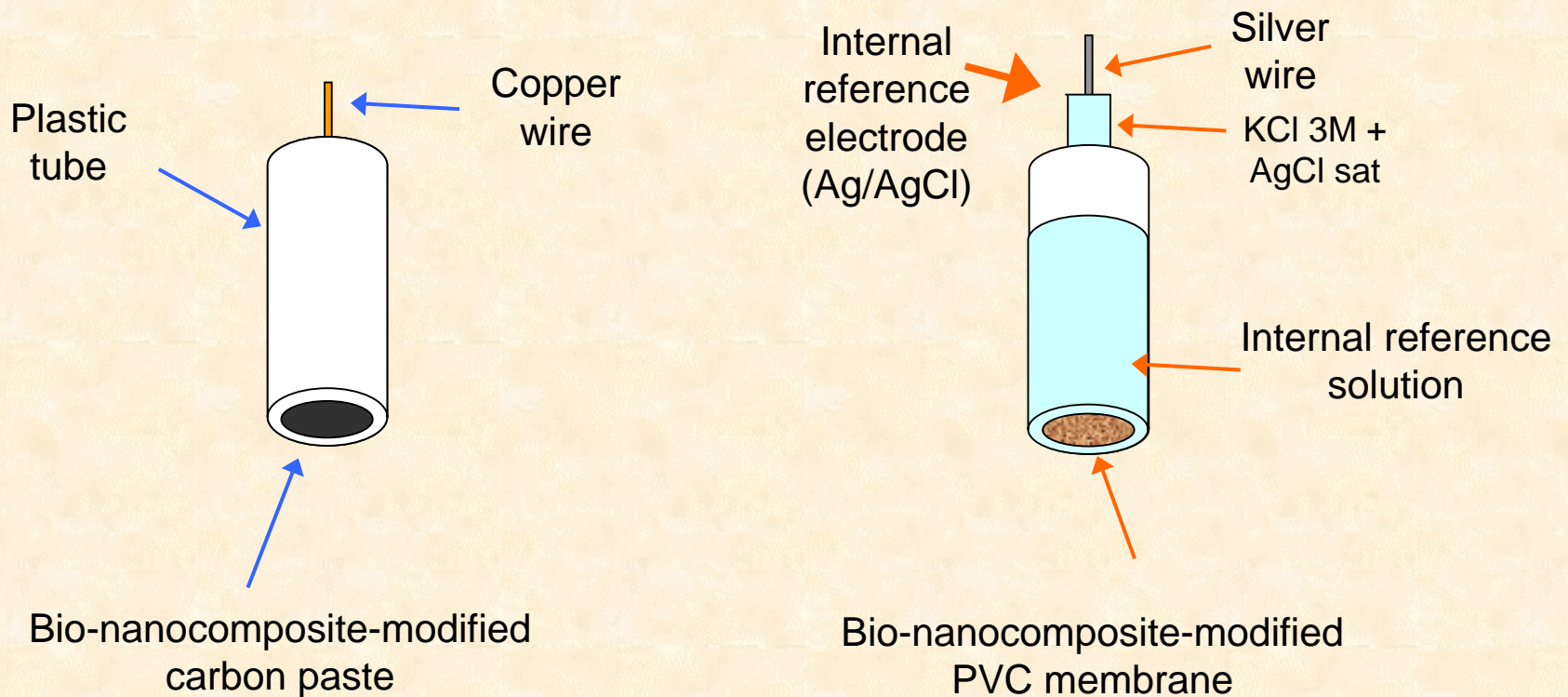
- Anion-exchange behavior

^{13}C NMR Spectroscopy

TREATMENT WITH NaNO_3



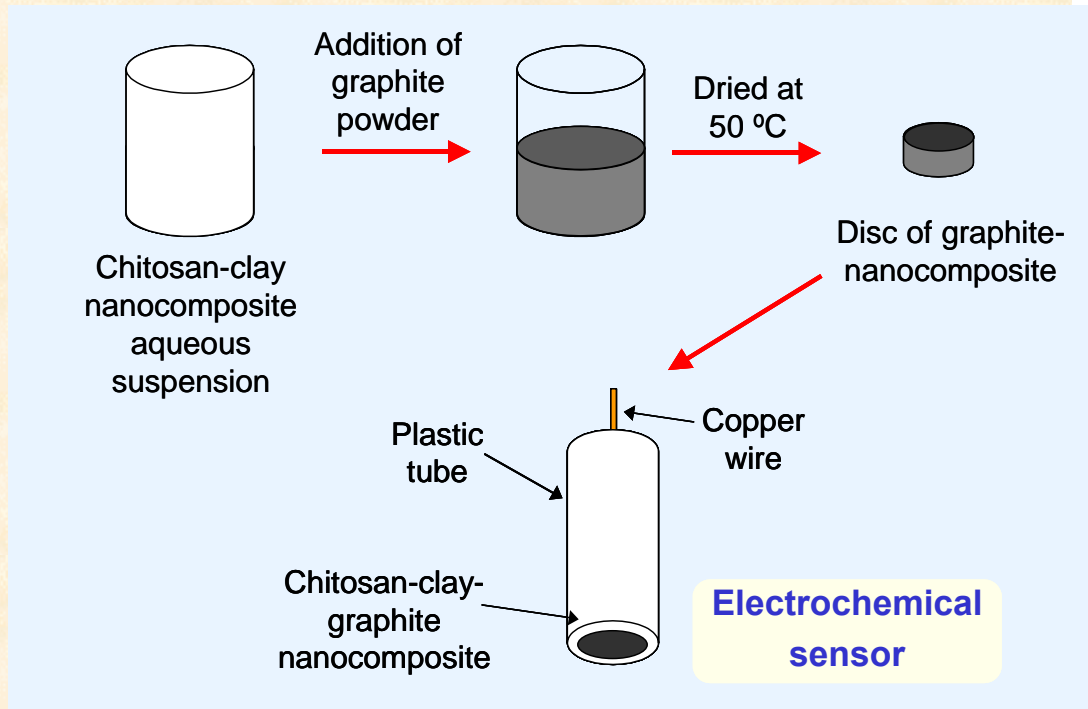
Development of potentiometric sensors



**Carbon paste electrodes
(CPEs)**

**PVC-membrane
based electrodes**

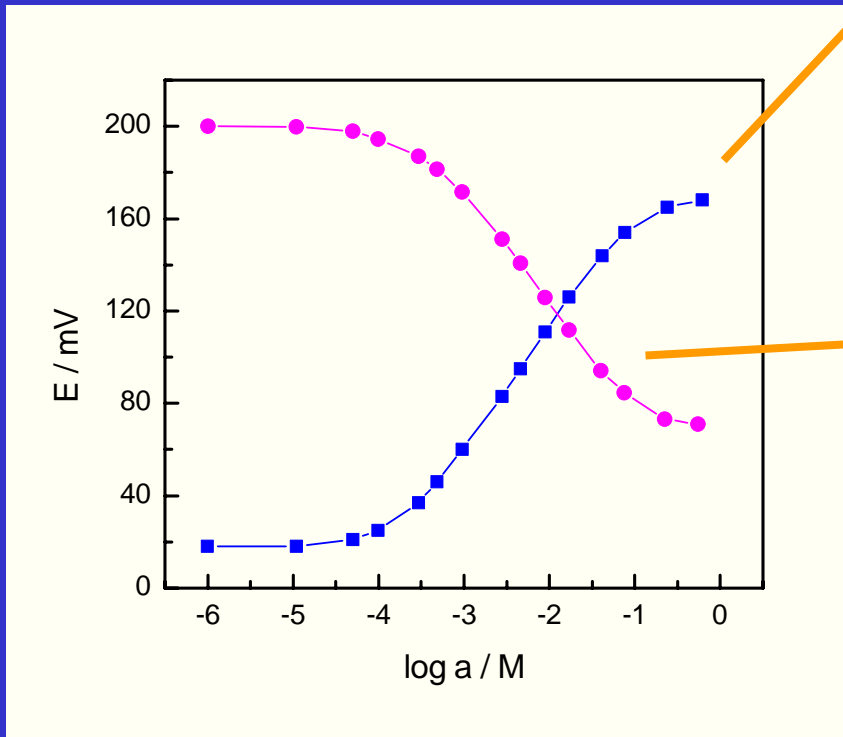
Chitosan-clay sensors development



The good mechanical properties of bio-nanocomposite avoid the use of binders as it is usual in CPEs or Epoxy Based Sensors

• Potentiometric response

Potentiometric measurement of NaCl with sensors prepared from different nanocomposites



Nanocomposite with a chitosan **monolayer**

Positive slope

Potentiometric response to **cations**

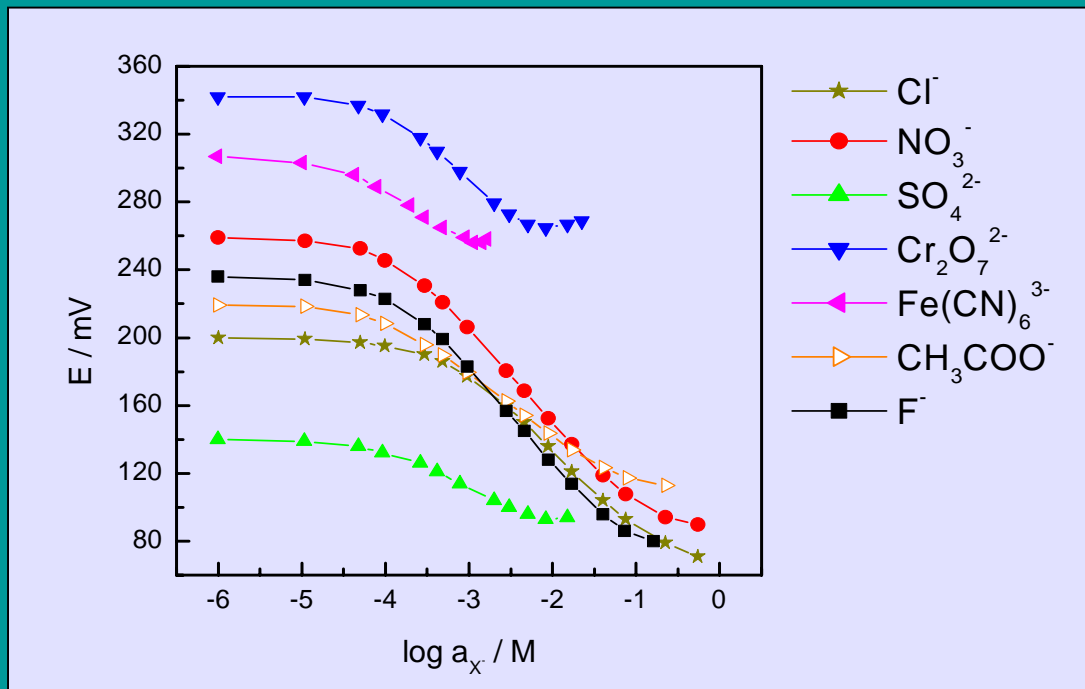
Nanocomposite with a chitosan **bilayer**

Negative slope

Potentiometric response to **anions**

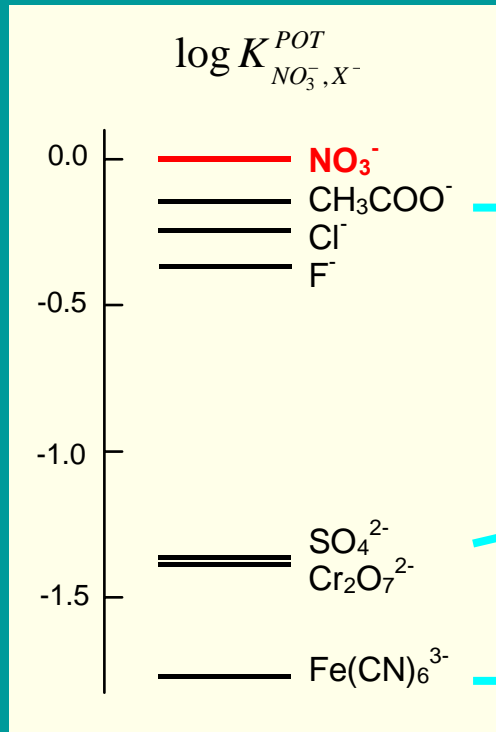
Chitosan-montmorillonite potentiometric sensor

The potentiometric response of a sensor based on the chitosan-clay nanocomposite with a chitosan **bilayer** is evaluated towards different anions



M. DARDER, M. COLILLA, E. RUIZ-HITZKY *Chitosan-clay nanocomposites: application as electrochemical sensors*
Appl. Clay Sci. **28**, 199-208 (2005)

Selectivity coefficients



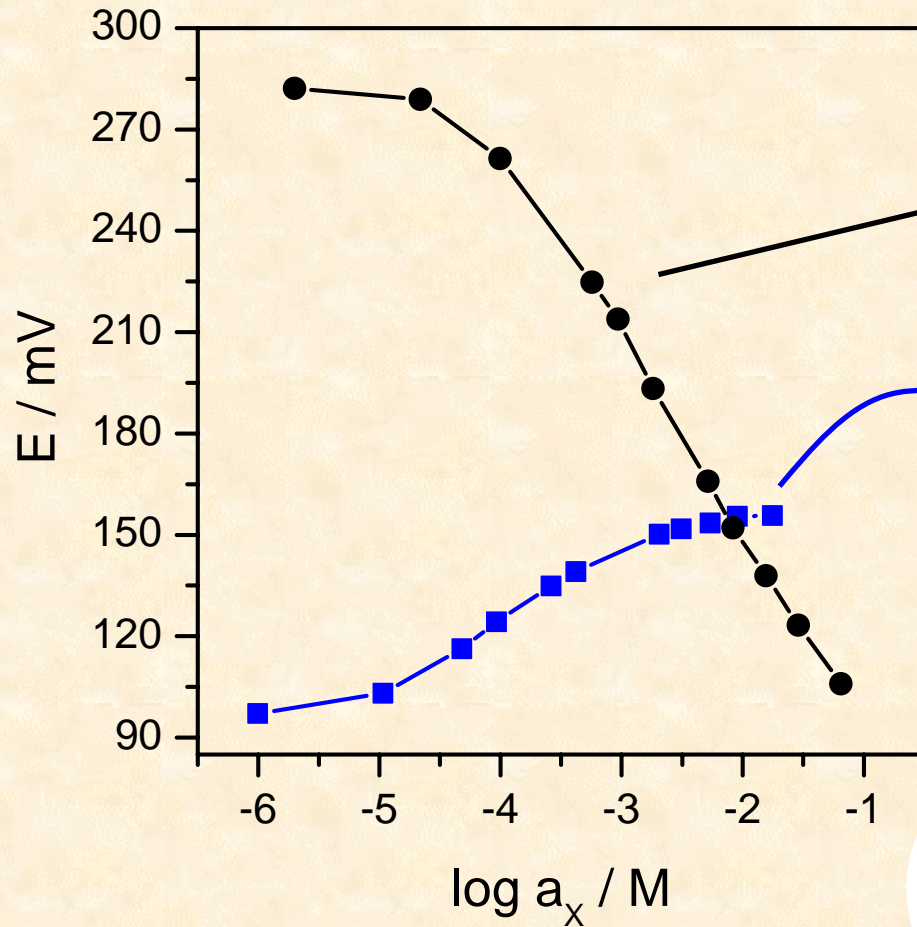
Partial selectivity
to **monovalent**
anions

Cross-sensitivity

The sensor is **25 times** more selective
for nitrate than **divalent anions**

The sensor is **60 times** more selective
for nitrate than **trivalent anions**

LDH Bio-nanocomposite CPEs



Zn₂Al/Cl LDH

$$E = 35.9 - 57.3 \log a_{\text{Cl}^-}$$
$$r = 0,9986$$

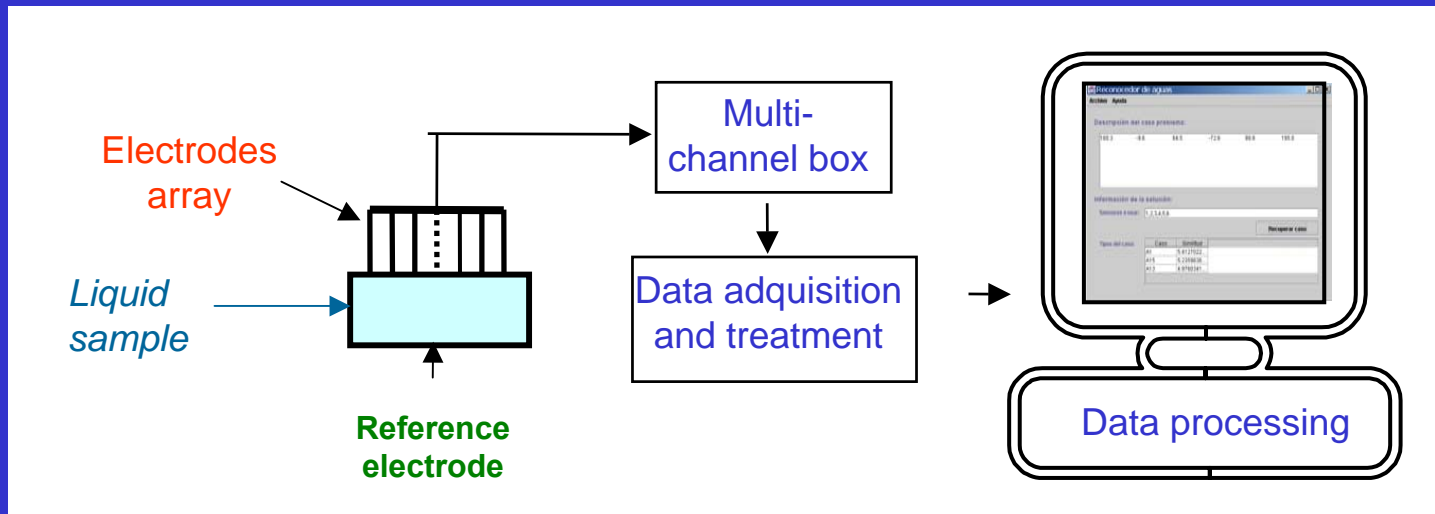
i-carrageenan-LDH

$$E = 213.5 + 22.2 \log a_{\text{Ca}^{2+}}$$
$$r = 0,998$$

The AEC of the LDH has been reversed to a CEC in the bionanocomposite

Sensor arrays controlled by Artificial Intelligence

Combination of sensors with AI techniques give sophisticated devices, as artificial noses and tongues



Application of **Case-Based Reasoning (CBR)** for multicomponent analysis

M. COLILLA, C.J. FERNÁNDEZ, E. RUIZ-HITZKY
The Analyst, 127, 1580-1582 (2002)

Applications

-water quality: drink water & industrial pollution

-Ions in biological fluids

-Automation in fertiirrigation

Home-made “electronic tongue” entirely developed in our lab, using sensors based on bio-nanocomposites, applied to control the water quality



CONCLUSIONS

*Cationic or Anionic polysaccharides form bio-nanocomposites by combination at the nanometric scale with layered silicates (INTERCALATION) or double hydroxides (CO-ORGANISED ASSEMBLY) resulting in **functional hybrid nanostructured materials** with:*

- *Good mechanical properties*
- *Controlled ion-exchange behaviour*

These results open a way to prepare novel bio-nanohybrids provided of structural or functional properties.

FUTURE WORK

The use of other layered host solids (phosphates, phosphonates, chalcogenides, etc.) & bio-polymers (polypeptides, nucleic acids, etc..)

Projects

MAT2003-06003-C02-01

07N / 0070 / 2002

IFAPA – 2002.000890

Financed by

CICYT

Comunidad Autónoma de Madrid

Junta de Andalucía